

Atmos'fair 2024

Development and validation of an on-line formaldehyde analyser for long-term air monitoring

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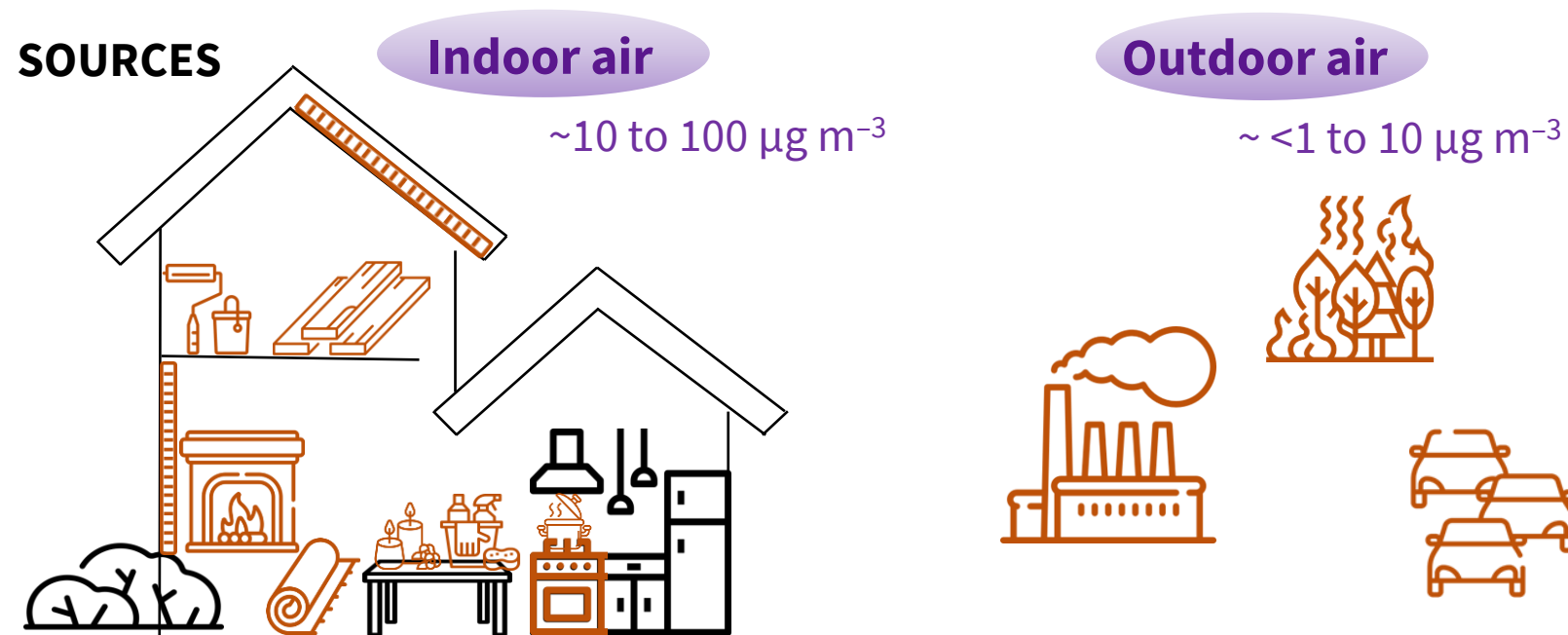
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Formaldehyde sources

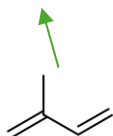
FORMALDEHYDE	
Structure	$\text{H}-\overset{\text{O}}{\parallel}{\text{C}}-\text{H}$
M (g mol ⁻¹)	30.03
Bp (°C)	-21°C
Henry Law's constant (atm ⁻¹)	2.5 10 ³
Odour threshold (µg m ⁻³)	50–500

SOURCES



In-situ formation:

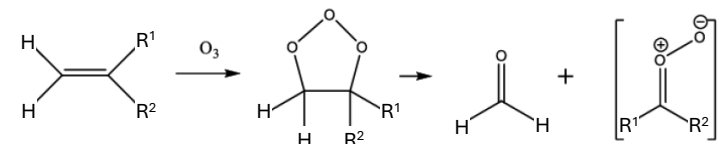
VOCs (alkanes or alkenes) oxidation by radicals



Methane, isoprene, most abundant hydrocarbon emission into the atmosphere

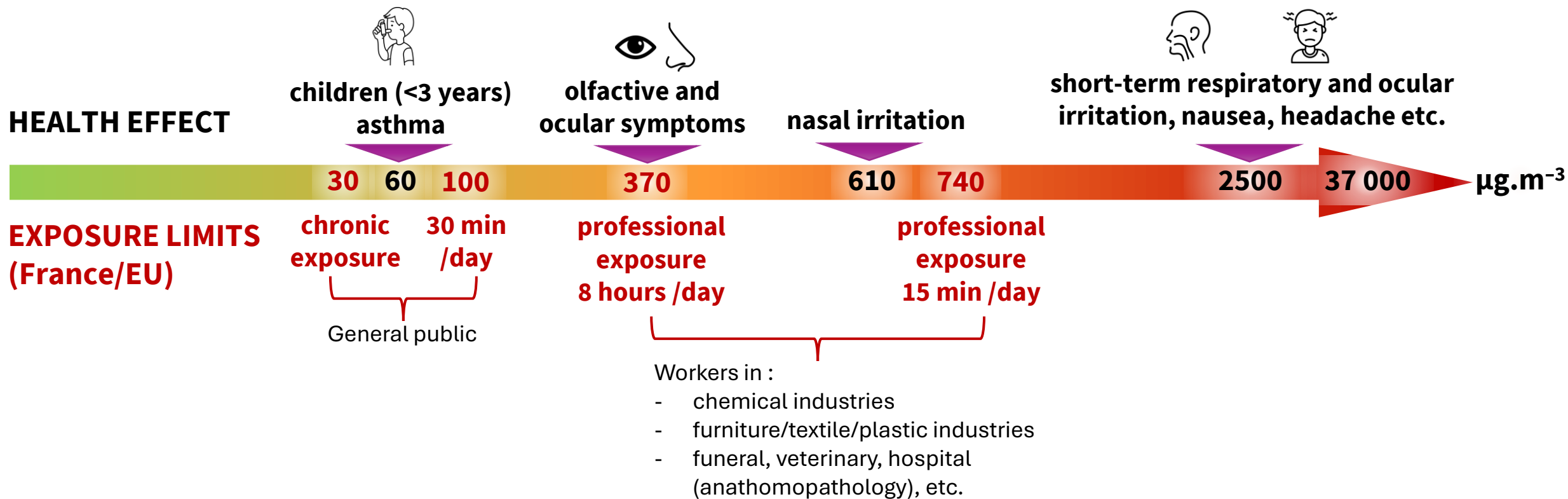
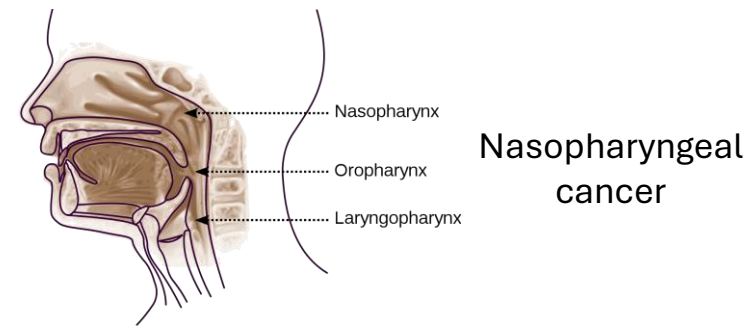
ex: terpenes used as fragrance in consumer products

alkene ozonation



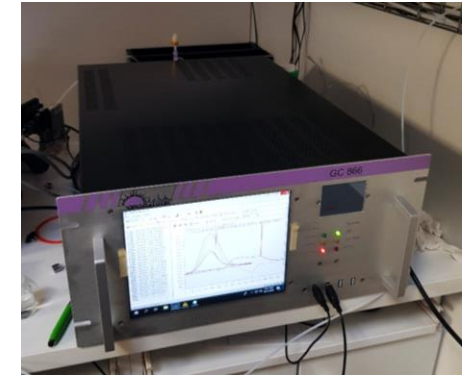
Salthammer, T., *Building and Environment*, 2019

Formaldehyde health effect and legislation



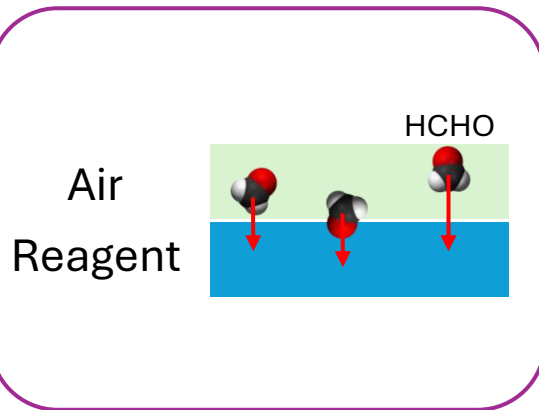
Analytical technique		Detection range ($\mu\text{g m}^{-3}$)	LOD ($\mu\text{g m}^{-3}$)	Temporal resolution	Weight (kg)	Price (k€)
IR/UV Optical spectroscopy	FTIR	0 – 12 300	1.2	< 1 min	20	≈ 55
	CRDS	0 – 37 000	0.2	< 0.03 – 5 min	21	≈ 70 – 80
Chemical method coupled to spectroscopy	Hantzsch fluorimetry	0 – 1230 0 – 3690	1.2 0.2	1 – 120 s 90 s	6.5 20	≈ 20 ≈ 35
	Separation techniques off-line DNPH HPLC-UV ISO 16000-3	0 – 15 000	0.6	hours/days	> 40 kg	≈ 70
Mass spectrometry	SIFT-MS	0 – 5000	0.5 – 9	<2 s	220	≈ 300
Electrochemistry	sensors	0 – 2 500 000	6 – 12	< 30 s	2 – 3	≈ 6

Principle of the on-line formaldehyde analyser: airmoF

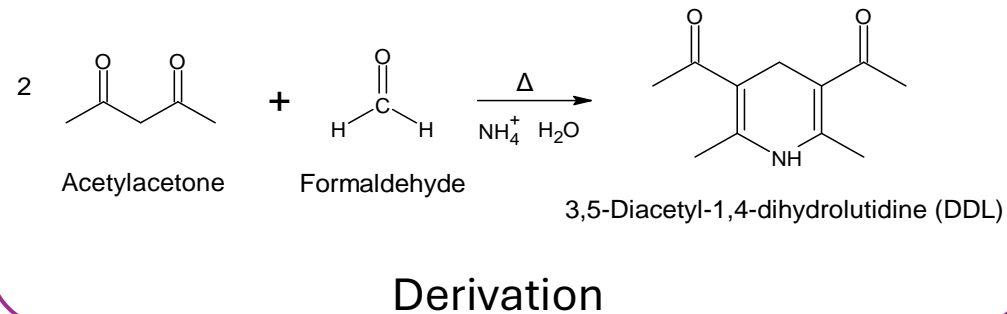


- Sampling by transfer of formaldehyde from the air to the liquid reagent
- **Derivation reaction** forming 3,5-diacetyl-1,4-dihydrolutidine (DDL)
- DDL **fluorescence** detection

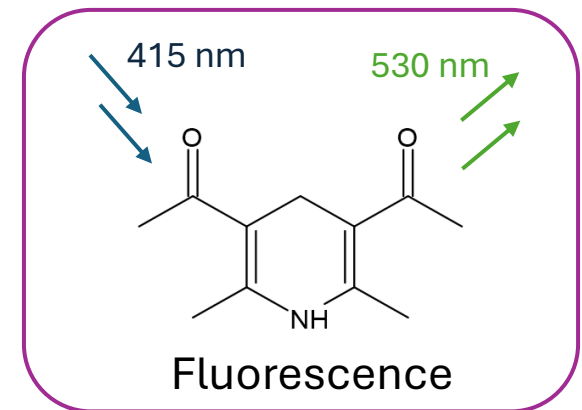
1. Uptake



2. Reaction



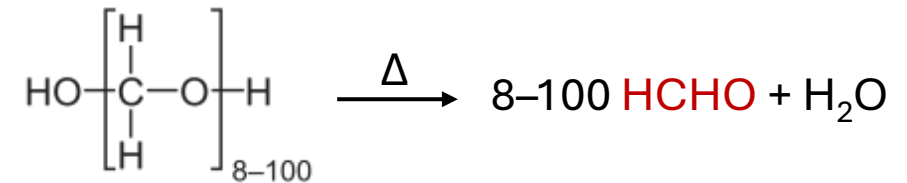
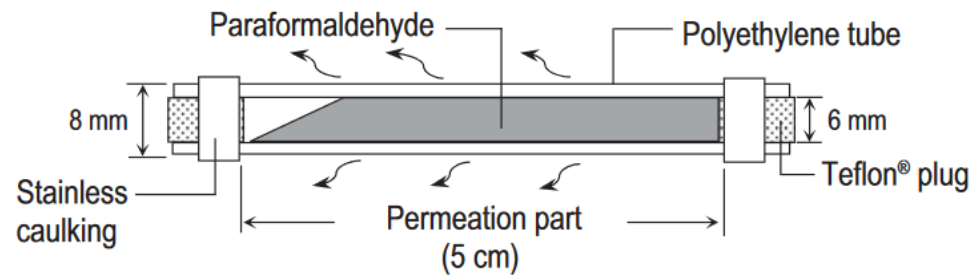
3. Detection



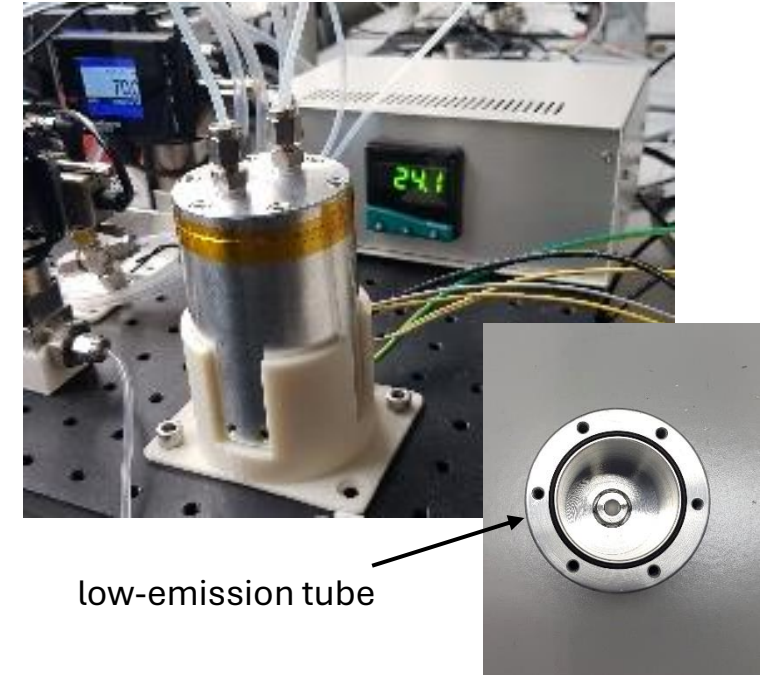
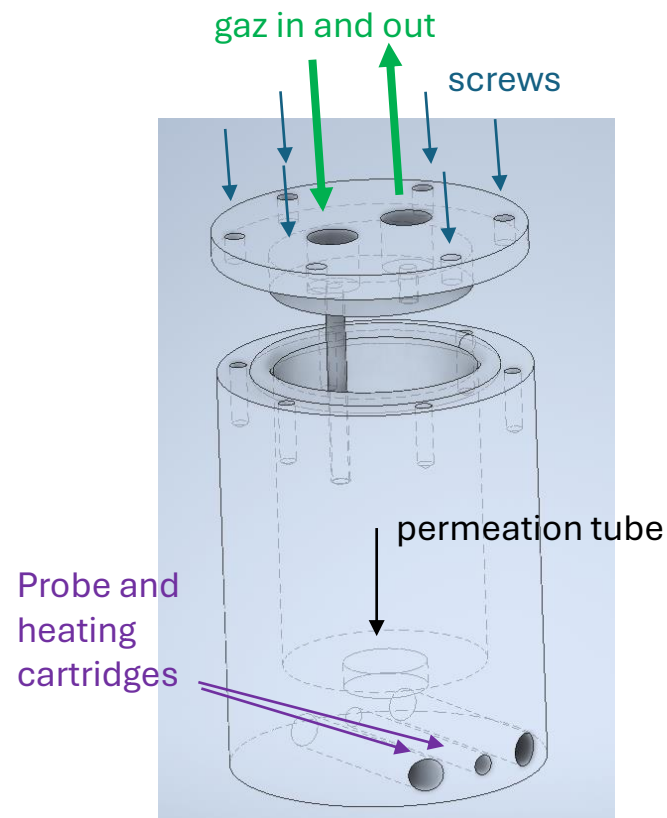
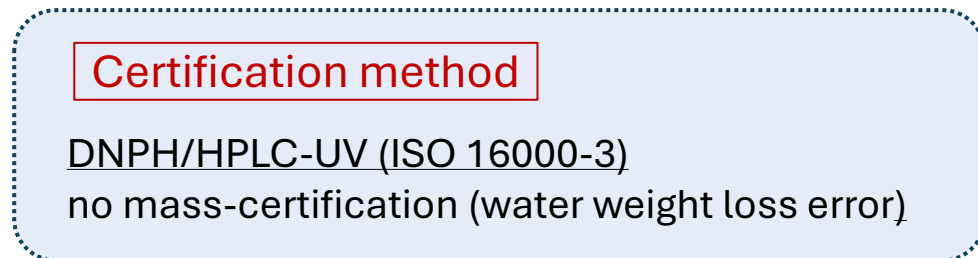
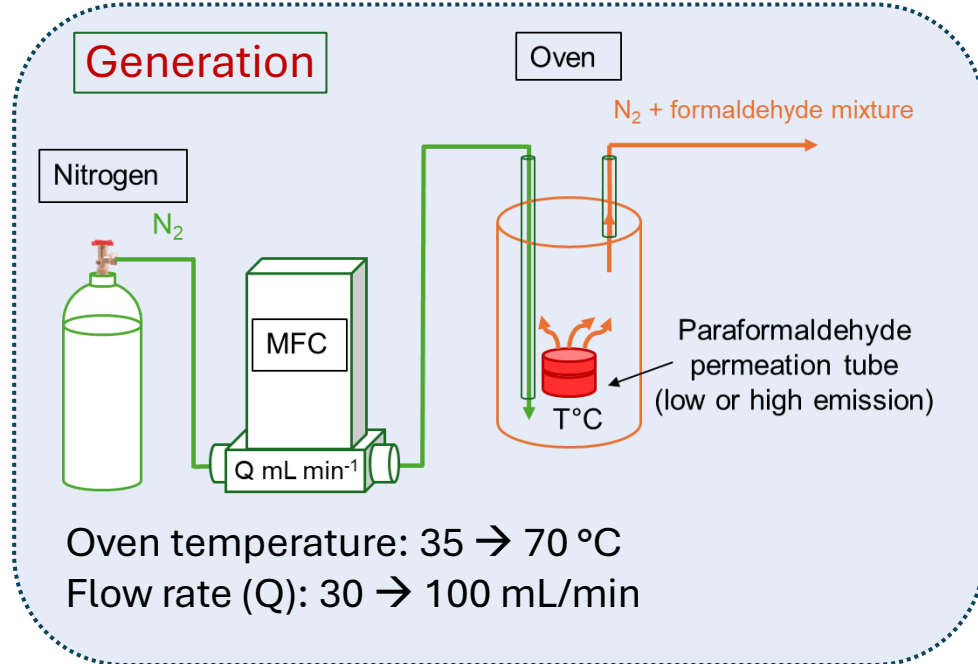
- Formaldehyde-specific
- On-line

Formaldehyde gas standard: permeation tube

- Mechanisms: **paraformaldehyde depolymerisation** and **permeation** through membrane

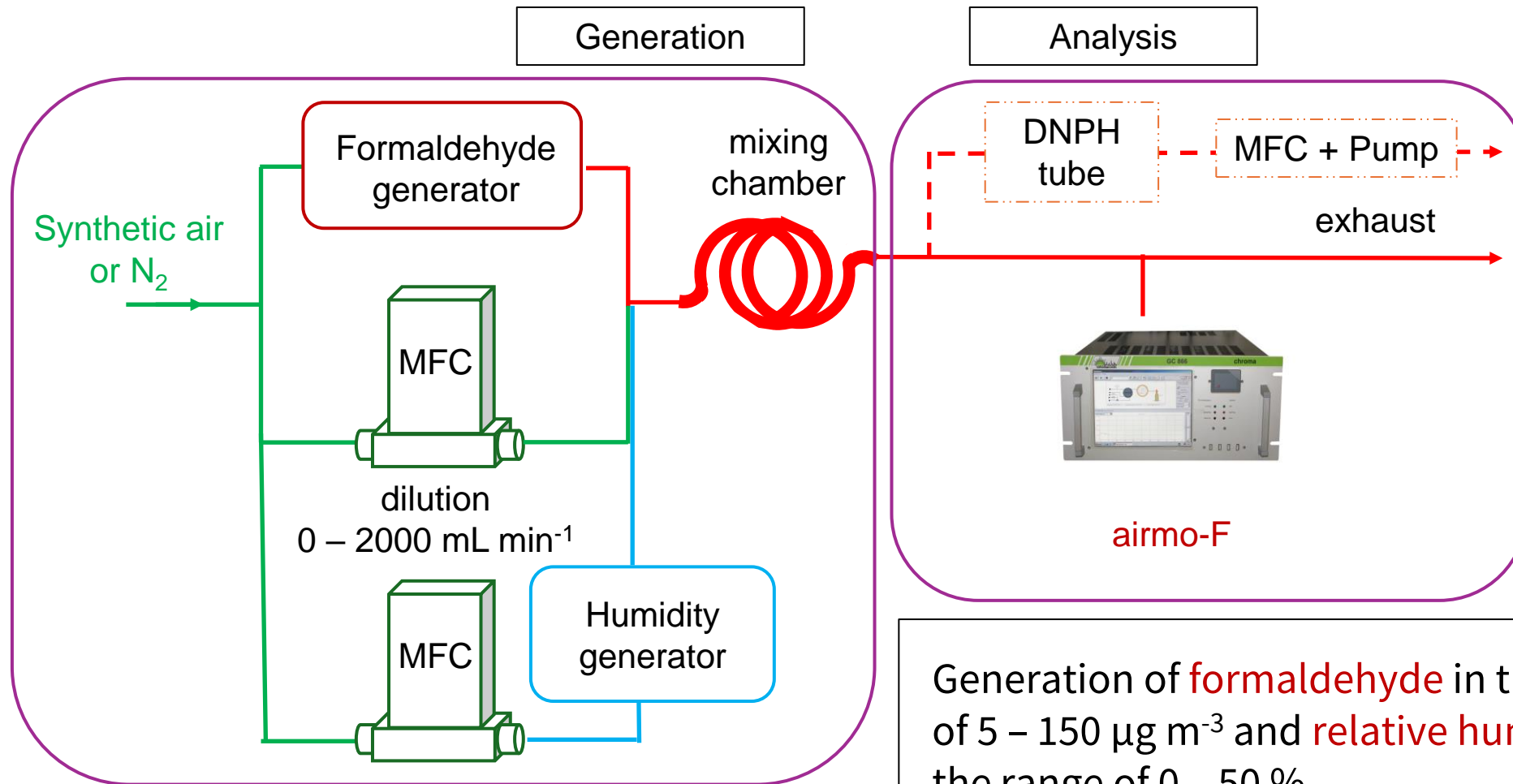


Formaldehyde gas standard: permeation tube



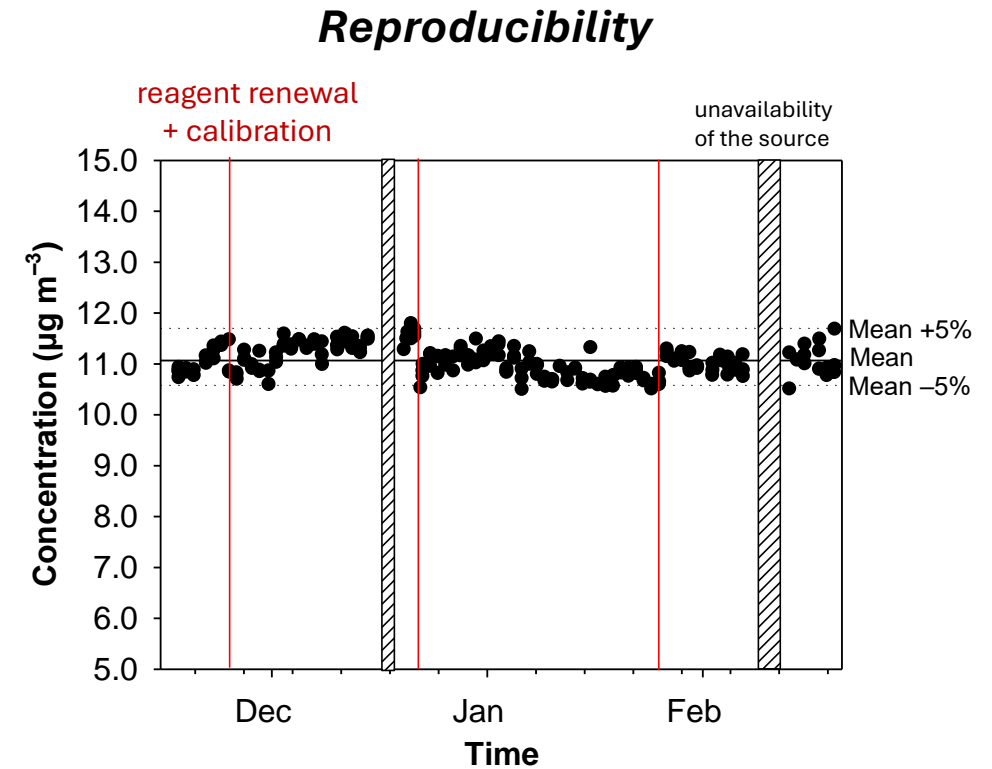
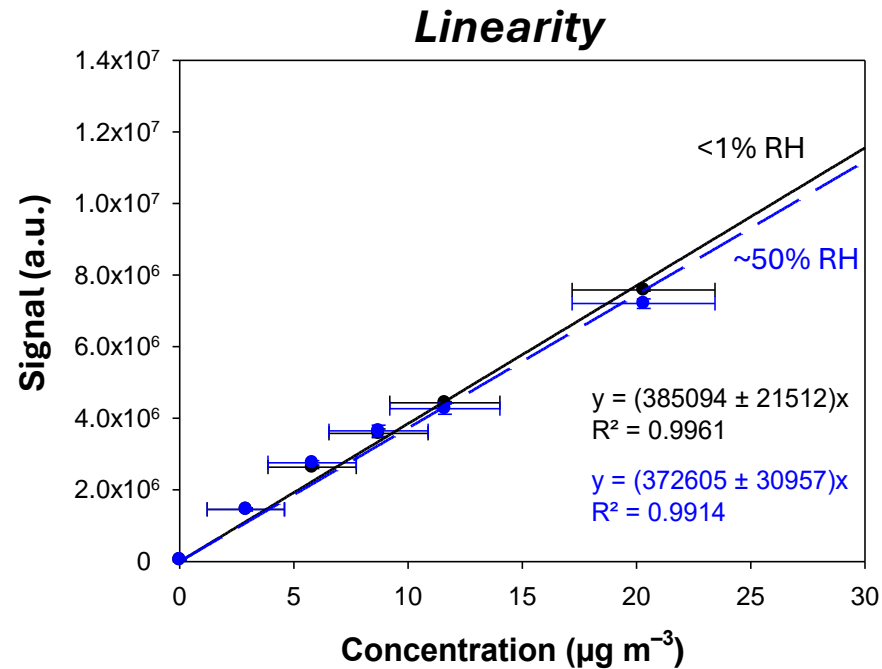
$\rightarrow 8.3 \text{ to } 464 \mu\text{g m}^{-3}$

Laboratory experiments set-up



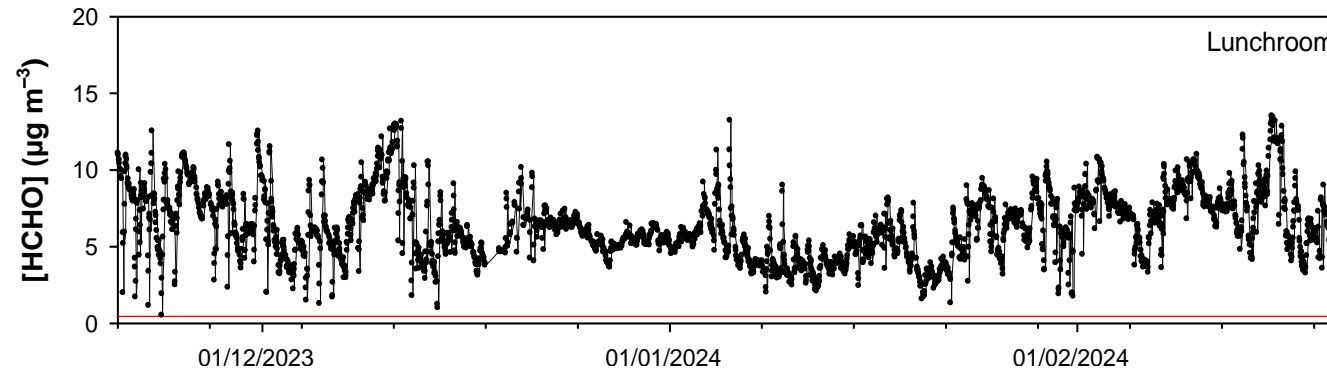
Generation of **formaldehyde** in the range of 5 – 150 $\mu g\ m^{-3}$ and **relative humidity** in the range of 0 – 50 %

Method validation

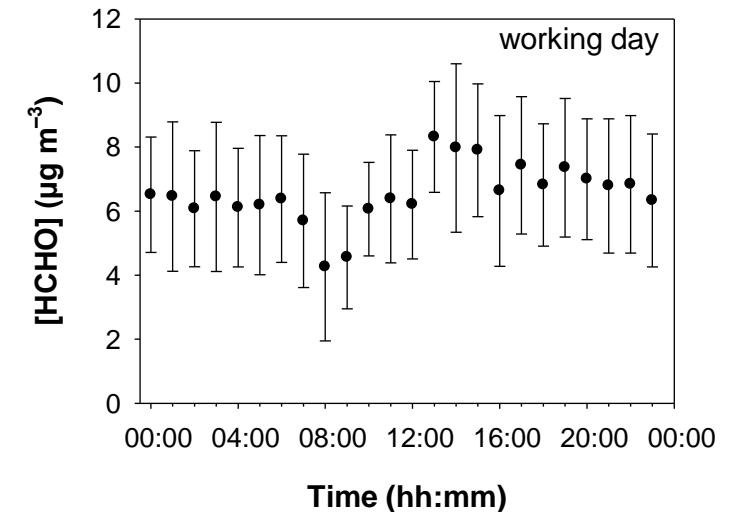
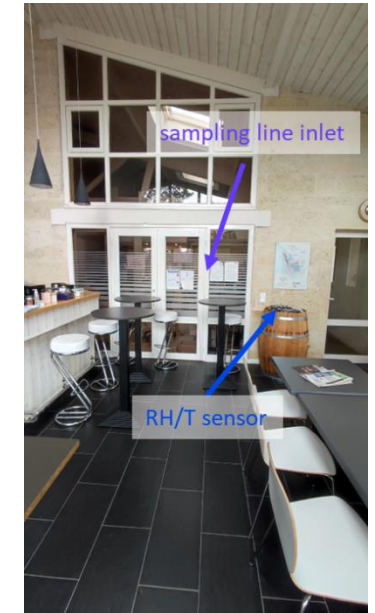
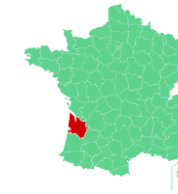


- Linearity: $R^2 > 0.99$ in humid and dry conditions
- Repeatability at $11 \mu\text{g m}^{-3}$ ($\text{RSD}_{n=10}$): 1.9%
- Reproducibility at $11 \mu\text{g m}^{-3}$ ($\text{RSD}_{3 \text{ months}}$): 2.6 %
- $\text{LOD}_{\text{US EPA}; <1\% \text{RH}} = 0.19 \mu\text{g m}^{-3}$

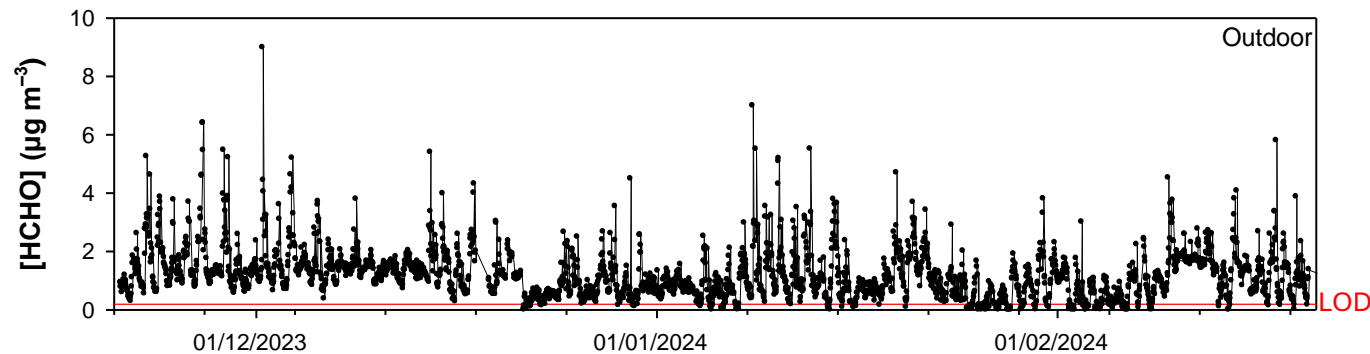
Field application – indoor, winter



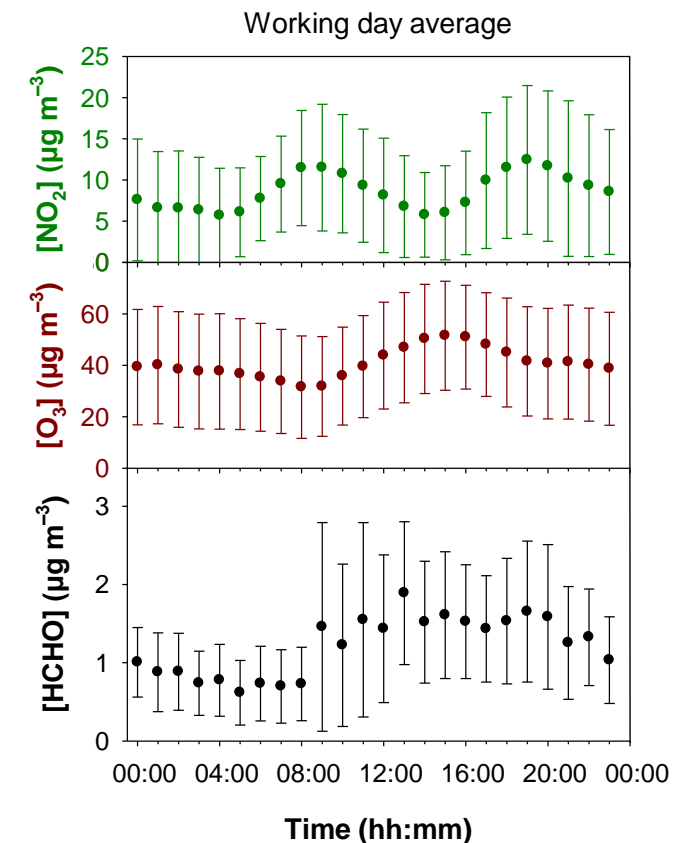
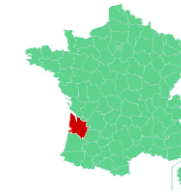
- Stable concentration during vacation
- 8AM: manual ventilation \rightarrow concentration decrease
- 1PM: lunchtime, human activity \rightarrow concentration increase



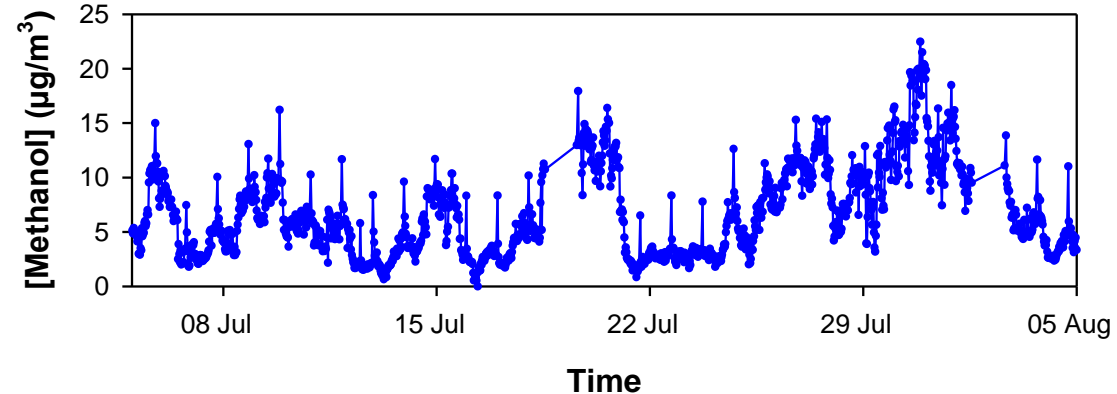
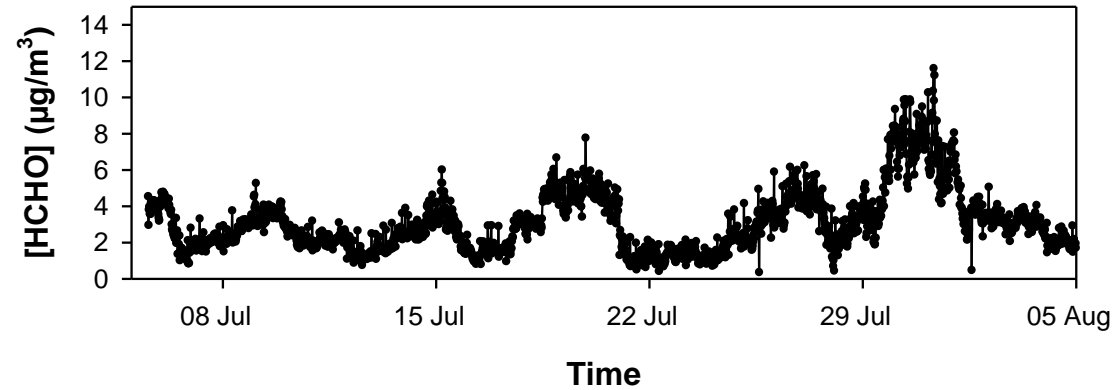
Field application – rural, winter



- **3-month outdoor air measurement:** rural area near a highway
- acquisition frequency enough to observe **daily variation**
- Diurnal variation correlated with ozone concentration \rightarrow in-situ formation
- Disturbed trend at commute hours \rightarrow primary emissions + in-situ formation



Field application – remote, summer



- Most measurements mostly between 1 and $5 \mu\text{g m}^{-3}$
- no diurnal trend observed
- correlation with methanol

Conclusion

- Sensitivity ✓
- Selectivity ✓
- Repeatability ✓
- Reproducibility ✓
- Low maintenance ✓
- Robustness ✓
- Fully automatic ✓
- **Field applicability** ✓
- Portable version : microF with LOD at $1\mu\text{g}/\text{m}^3$ in automatic
 - →adapted for indoor air study

LOD $< \mu\text{g m}^{-3} \rightarrow 0.19 \mu\text{g m}^{-3}$, suitable for outdoor air levels

formaldehyde-specific, **no water interference**

$< 2\%$ at $11 \mu\text{g m}^{-3}$ (RSDn=10)

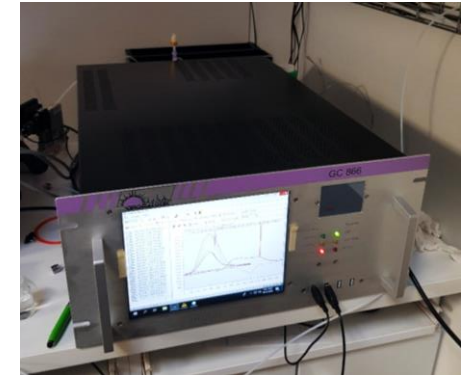
reproducible results over 3–4 months

monthly reagent renewal

peak integration

rural, remote, indoor

airmoF



microF





anaVOC
LABCOM



Thank you for your attention

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